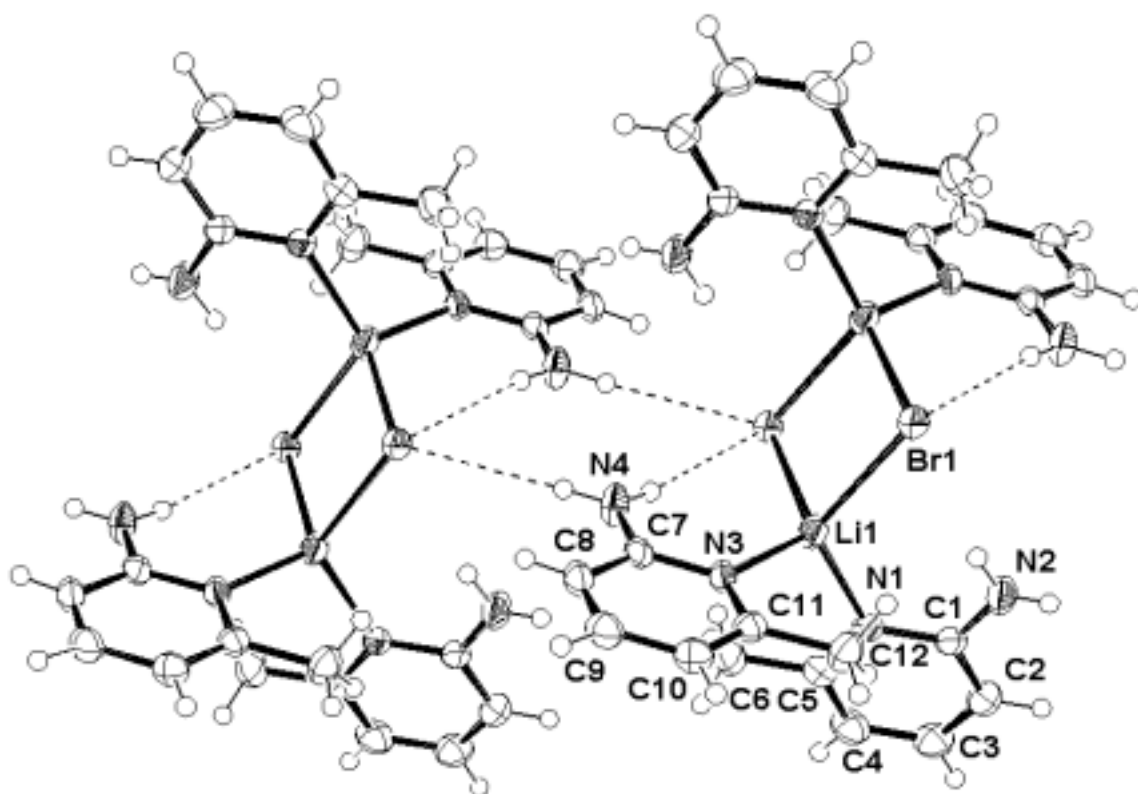


**THE MOLECULAR STRUCTURE OF**  
 **$[(C_5H_3NMeNH_2)_2Li(\mu-Br)_2Li[(C_5H_3NMeNH_2)_2]]_n$** ,  
 **$(C_5H_3NMeNH_2 = 6\text{-methyl-2-aminopyridine})$**

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**Figure 1.** Molecular structure of  $[(C_5H_3NMeNH_2)_2Li(\mu-Br)_2Li[(C_5H_3NMeNH_2)_2]]_n$ . Selected bond lengths (Å) and angles ( $^\circ$ ): Li1-Br1 2.614(5), Li1-Br1#1 2.616(5), Li1-N1 2.116(5), Li1-N3 2.100(5), Li1...Li1#1 3.544(5), Br1...Br1#1 3.846(5), H2NA-Br1 2.54(5), H4A-Br1#1 2.65(5), H4B-Br1#2 2.76(5), N2-Br1 3.469(3), N4-Br1#1 3.504(2), N4-Br1#2 3.632(2), N1-Li1-Br1 119.20(19), N3-Li1-Br1 105.65(19), N1-Li1-Br1#1 103.90(19), N3-Li1-Br1#1 119.98(19), N1-Li1-N3 112.9(2), Li1-Br1-Li1#1 85.32(15), Br1-Li1-Br1#1 94.68(15). Symmetry transformations used to generate equivalent atoms: #1  $-x+1, -y+2, -z+2$ , #2  $x+1, y, z$

#### Comment

The 2:1 adduct of 6-methyl-2-aminopyridine and LiBr crystallizes as a centrosymmetric dimer with doubly bridging bromides. This structure is reminiscent of the related 2:1 complexes formed between 2-methylpyridine and LiCl, LiBr or LiI.[1] Perhaps surprisingly, the present complex is not isostructural with the isoelectronic adducts of 2,6-dimethylpyridine and LiBr where a 1.5:1 complex crystallizes as a tetranuclear ladder type structure.[2] The over-riding influence affecting the overall structure in the present dimer may be the inter- and intra-molecular hydrogen bonds formed between N-H donors and Br acceptors forming polymeric strands between adjacent dimers. The intramolecular hydrogen bond formed between

N4-H and Br1' does not significantly affect the Li-N and Li-Br bond lengths within the  $Li_2Br_2$  core where the Li-Br bond lengths are almost identical, a feature that contrasts the 2:1 adducts of 2-methylpyridine and LiX.[1]

### Experimental

#### Preparation of $[(C_5H_3NMeNH_2)_2Li(\mu-Br)_2Li(C_5H_3NMeNH_2)_2]$ :

To anhydrous LiBr (0.33g, 3.8 mmol) in thf (20 cm<sup>3</sup>) was added to a solution of 6-methyl-2-aminopyridine (0.80 g, 7.2 mmol) in thf (20 cm<sup>3</sup>) and the solution stirred for 2h. The solution was reduced in volume to the point of crystallization and placed at -35°C overnight, yielding colourless crystals. Yield 0.90 g, 79%, mp 63-5°C. <sup>1</sup>H-NMR (400MHz, CDCl<sub>3</sub>, 298K)  $\delta$  2.30 (s, 12H, CH<sub>3</sub>), 4.37 (s, 8H, NH<sub>2</sub>), 6.22 - 7.25 (m, 12H, H<sub>arom</sub>) ppm. <sup>13</sup>C-NMR (100.6 MHz, CDCl<sub>3</sub>, 298K)  $\delta$  25.98 (CH<sub>3</sub>), 105.77, 113.60, 138.47, 157.21, 158.27 (aromatic C) ppm. IR (Nujol, NaCl plates,  $\nu$ ): 3220, 2971, 2360, 1348, 1281, 1164, 1049, 993, 894, 784 cm<sup>-1</sup>.

#### Crystallography:

**Table 1.** Crystal data for of  $[(C_5H_3NMeNH_2)_2Li(\mu-Br)_2Li(C_5H_3NMeNH_2)_2]$

Formula	C <sub>24</sub> H <sub>32</sub> Br <sub>2</sub> Li <sub>2</sub> N <sub>8</sub>	Formula weight	606.28
Crystal system	triclinic	Crystal size, mm	0.30 x 0.24 x 0.24
Space Group	<i>P</i> bar1	<i>a</i> , Å	8.2173(6)
<i>b</i> , Å	8.5030(6)	<i>c</i> , Å	11.5074(8)
$\alpha$ , °	92.635(2)	$\beta$ , °	108.650(1)
$\gamma$ , °	112.086(1)	<i>V</i> , Å <sup>3</sup>	693.15(9)
<i>Z</i>	1	Diffractionmeter	Siemens SMART 1000
Temperature, K	223(2)	$\mu$ (Mo-K $\alpha$ ), mm <sup>-1</sup>	2.951
<i>D</i> <sub>calcd</sub> , g cm <sup>-3</sup>	1.452	<i>F</i> (000)	308
$\theta$ <sub>max</sub> , °	27.07	Reflns meas.	3719
Reflns unique	2642	Reflns with <i>I</i> > 2 $\sigma$ ( <i>I</i> )	2333
<i>R</i> ( <i>F</i> <sup>2</sup> ) [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )]	0.030	<i>R</i> <sub>w</sub> ( <i>F</i> <sup>2</sup> ) (all data)	0.078
$\rho$ , e Å <sup>-3</sup>	0.49	G.O.F	1.01
No. parameters	209		
Weighting scheme	w=1/[ $\sigma^2(F_o^2)+(0.0491P)^2+0.0966P$ ] where P=( <i>F</i> <sub>o</sub> <sup>2</sup> +2 <i>F</i> <sub>c</sub> <sup>2</sup> )/3		
Programs used	SHELX-97 [3], XSEED [4], SADABS [5]		
Deposition number	CCDC 168987		

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