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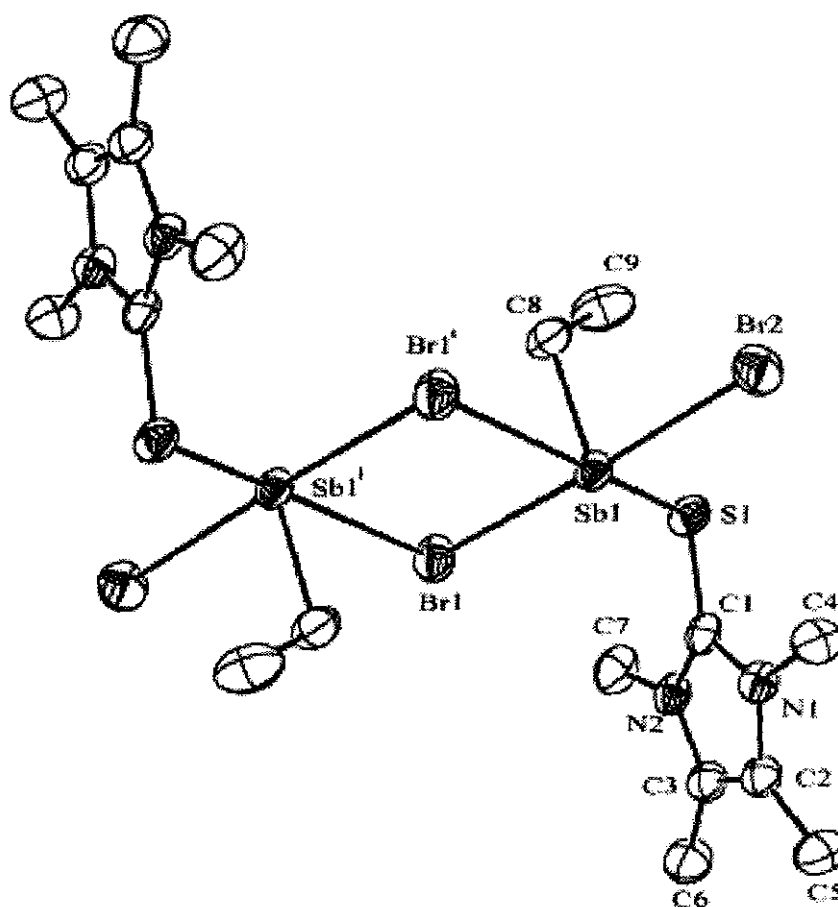
# THE MOLECULAR STRUCTURE OF $\{[SbEtBr(\mu-Br)[SCN(Me)C_2Me_2N(Me)]]_2\}$

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**Figure 1.** Molecular structure of  $\{[SbEtBr(\mu-Br)[SCN(Me)C_2Me_2N(Me)]]_2\}$ . Selected bond lengths (Å) and angles( $^{\circ}$ ): Sb(1)-C(8) 2.166(5), Sb(1)-S(1) 2.531(1), Sb(1)-Br(2) 2.7286(9), Sb(1)-Br(1) 2.8554(9), Sb(1)-Br(1') 3.227(1), S(1)-C(1) 1.721(5), N(1)-C(1) 1.342(6), N(2)-C(1) 1.335(6), N(1)-C(2) 1.390(6), N(2)-C(3) 1.387(6), C(2)-C(3) 1.353(7), C(8)-Sb(1)-S(1) 91.5(2), C(8)-Sb(1)-Br(2) 86.7(2), S(1)-Sb(1)-Br(2) 88.14(4), C(8)-Sb(1)-Br(1) 87.8(2), S(1)-Sb(1)-Br(1) 87.41(4), Br(2)-Sb(1)-Br(1) 172.81(2), Sb(1)-Br(1)-Sb(1') 93.21(4), C(1)-S(1)-Sb(1) 102.9(2). Symmetry operator used to generate equivalent atoms: 1-x, 1-y, -z.

## Comment

Prior to this work there were no crystallographically characterised examples of imidazolethione adducts of organoantimony compounds and only 3 examples of imidazolethione adducts of antimony trihalides [1-3]. The antimony centre in the present compound has a slightly distorted square based pyramidal geometry and a stereochemically active lone pair. The compound sits on an inversion centre and is dimeric through unsymmetrical Sb-Br-Sb bridges, both the longer and shorter interactions of which lie in the normal region for Sb-Br bonds [4]. The geometry of the imidazolethione ligand in the complex is similar to its geometry in the uncoordinated state [5].

## Experimental

### Preparation of $[\{\text{SbEtBr}(\mu\text{-Br})\{\text{SCN}(\text{Me})\text{C}_2\text{Me}_2\text{N}(\text{Me})\}\}_2]$ :

A solution of  $\text{SCN}(\text{Me})\text{C}_2\text{Me}_2\text{N}(\text{Me})$  (1.0 g, 6.4 mmol) in THF (20 mL) was added over 15 minutes to a solution of  $\text{SbEtBr}_2$  (2.0 g, 6.4 mmol) in THF (40 mL) at  $-50\text{ }^\circ\text{C}$ . The resulting yellow solution was warmed to room temperature and stirred overnight. Volatiles were removed *in vacuo* and the residue recrystallised from  $\text{CH}_2\text{Cl}_2$  to give the title compound as yellow prisms (2.15 g, 72%); m.p.  $123\text{--}126\text{ }^\circ\text{C}$ ;  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ , 300 K):  $\delta$  1.65 (6H, br.,  $\text{SbCH}_2\text{CH}_3$ ), 2.11 (12H, s, NCMe), 2.36 (4H, br.,  $\text{SbCH}_2$ ), 3.45 (12H, s, NMe);  $^{13}\text{C}$  NMR (100.6 MHz,  $\text{C}_6\text{D}_6$ , 300 K):  $\delta$  9.1 (s,  $\text{NCCH}_3$ ), 12.2 (s,  $\text{SbCH}_2\text{CH}_3$ ), 27.3 (s,  $\text{SbCH}_2$ ), 33.2 (s, NMe), 122.7 (s,  $\text{CCH}_3$ ), 163.4 (s, CS); MS APCI  $m/z$  (%) 467 [ $\text{M}^+$ , 17], 156 [ $\{\text{CN}(\text{Me})\text{C}_2\text{Me}_2\text{N}(\text{Me})\}^+$ , 100]; IR (Nujol) $\nu/\text{cm}^{-1}$ : 1550 m, 1242 s, 115 m, 872 s.

### Crystallography:

**Table 1.** Crystal data for  $[\{\text{SbEtBr}(\mu\text{-Br})\{\text{SCN}(\text{Me})\text{C}_2\text{Me}_2\text{N}(\text{Me})\}\}_2]$

Formula	$\text{C}_{18}\text{H}_{34}\text{Br}_4\text{N}_4\text{S}_2\text{Sb}_2$	Formula weight	933.76
Crystal system	triclinic	Crystal size, mm	0.64x0.50x0.28
Space Group	$P\bar{1}$	$a$ , Å	9.266(2)
$b$ , Å	9.536(3)	$c$ , Å	9.953(3)
$\alpha$ , °	113.96(1)	$\beta$ , °	100.049(18)
$\gamma$ , °	99.406(16)	$V$ , Å <sup>3</sup>	764.0(4)
$Z$	1	Diffractionmeter	CAD4
$\mu(\text{Mo-K}\alpha)$ , $\text{mm}^{-1}$	7.148	$D_{\text{calcd}}$ , $\text{g cm}^{-3}$	2.029
$F(000)$	888	$\theta_{\text{max}}$ , °	25.0
Reflns meas.	2862	Reflns unique	2681
Reflns with $I > 2\sigma(I)$	2390	$R(F^2)$ , $R_w(F^2)$ (all data)	0.037, 0.081
$\rho$ , $\text{e}\text{Å}^{-3}$	0.96	G.O.F	1.03
No. parameters	142	Absorp. correct. [6]	Difabs
Programs used	SHELX-97 [7], Ortep-3 [8]		
Deposition number	CCDC 165463		

### Acknowledgements

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### References

- [1] B. Rubin, F.J. Heldrich, W.K. Dean, D.J. Williams, A. Viehbeck, *Inorg. Chem.* (1981) **20** 4434.
- [2] P. Berges, W. Hinrichs, J. Kopf, D. Mandak, G. Klar, *J. Chem. Res.* (1985) **218** 2601.
- [3] D.J. Williams, D. Vanderveer, R.L. Jones, D.S. Menaldino, *Inorg. Chim. Acta* (1989) **165** 173.
- [4] A survey of the Cambridge Structural Database showed Sb-Br bond lengths lie in the range 2.444 Å – 3.596 Å (mean: 2.728 Å).
- [5] N. Kuhn, J. Fahl, R. Fawzi, M. Steinmann, *Z. Kristallogr.* (1998) **213** 434.
- [6] N.P.C. Walker, D. Stuart, *Acta Crystallogr. Sect. A* (1983) **39** 158.
- [7] G.M. Sheldrick, *SHELX-97*, University of Göttingen, Germany (1997).
- [8] L.J. Farrugia, *Ortep-3 for Windows*, University of Glasgow (1998).