

This is the **Accepted Version** of a paper published in the
journal: Main Group Metal Chemistry

Jones, Cameron, Junk, Peter C., and Cole, Marcus L. (2001) *The molecular structure of $[AlH_2Cl(Quinuclidine)_2]$* . Main Group Metal Chemistry, 24 (4). pp. 249-250.

<http://dx.doi.org/10.1515/MGMC.2001.24.4.249>

THE MOLECULAR STRUCTURE OF $[\text{AlH}_2\text{Cl}(\text{Quinuclidine})_2]$

Cameron Jones*¹, Peter C. Junk² and Marcus L. Cole¹

¹ Department of Chemistry, University of Wales, Cardiff, P.O. Box 912, Park Place, Cardiff, UK, CF10 3TB, jonesca6@cardiff.ac.uk

² Department of Chemistry, James Cook University, Townsville, Qld, 4811, Australia.

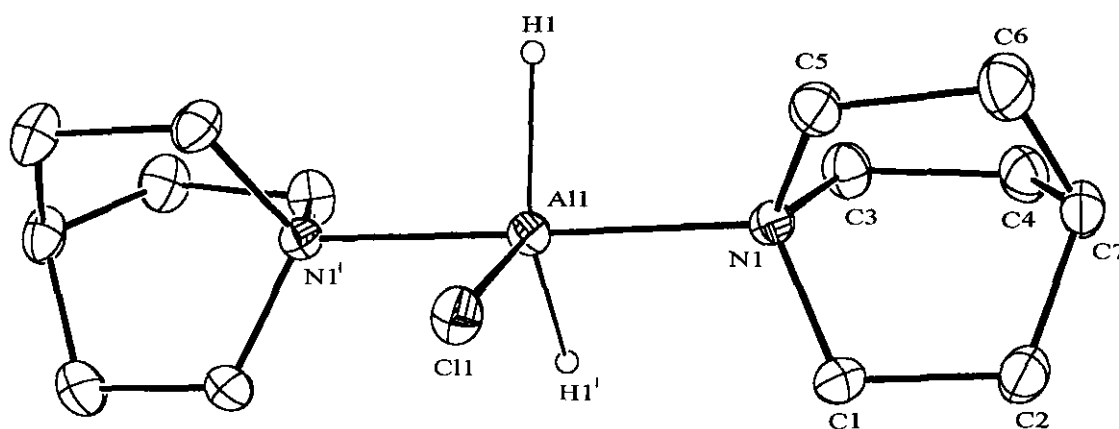


Figure 1. Molecular structure of $[\text{AlH}_2\text{Cl}(\text{Quinuclidine})_2]$. Selected bond lengths (Å) and angles(°): Cl(1)-Al(1) 2.2145(13), Al(1)-N(1) 2.1354(17), Al(1)-H(1) 1.63(2), N(1)-Al(1)-N(1') 175.90(10), N(1)-Al(1)-Cl(1) 92.05(5), N(1)-Al(1)-H(1) 88.9(8), N(1')-Al(1)-H(1) 89.3(8), H(1)-Al(1)-Cl(1) 116.2(7), H(1)-Al(1)-H(1') 127.7(9).

Comment

The hydride ligand H(1) was located from difference maps and refined isotropically. The Al(1)-H(1) distance is in the normal region for such interactions [1]. The compound is isomorphous to its gallane counterpart, $[\text{GaH}_2\text{Cl}(\text{Quinuclidine})_2]$ [2].

Experimental

Preparation of $[\text{AlH}_2\text{Cl}(\text{Quinuclidine})_2]$:

Solid quinuclidine hydrochloride (1.55g, 10.5 mmol) was added to a stirred slurry of LiAlH_4 (0.2g, 5.3 mmol) in diethyl ether (40 cm^3) sustained at a temperature of -50°C . The resultant colourless slurry was then warmed to room temperature over 5hrs, by which time gas evolution had ceased. Volatiles were removed *in vacuo* to yield an oily white solid that was subsequently recrystallised from diethyl ether. Placement at -30°C overnight gave the title compound as colourless prisms. Yield 1.19g, 79%; mp 182°C (dec). ^1H NMR (400 MHz, C_6D_6 , 300K): δ 1.12 (12H, m, CH_2), 1.24 (2H, m, CH), 2.64 (12H, t, ^3JHH 9.2 Hz, NCH_2), 4.22 (2H, br s, Al-H). ^{13}C NMR (100.6 MHz, C_6D_6 , 300K): δ 19.8 (s, CH), 27.6 (s, CH_2), 56.3 (s, NCH_2). MS EI: m/z (%) 176 [$\{\text{M-Quin}\}^+$, 100], 251 [$\{\text{M-Cl}\}^+$, 64], 111 [Quin^+ , 42]. IR (Nujol)/ cm^{-1} : 1691 cm^{-1} (s, br, Al-H).

Table 1. Crystal data for [AlH₂Cl(Quinuclidine)₂]

Formula	C ₁₄ H ₂₈ AlClN ₂	Formula weight	286.81
Crystal system	orthorhombic	Crystal size, mm	0.15x0.15x0.15
Space Group	<i>Pbcn</i>	<i>a</i> , Å	10.6225(10)
<i>b</i> , Å	12.1485(11)	<i>c</i> , Å	12.4366(11)
<i>V</i> , Å ³	1604.9(3)	<i>Z</i>	4
Diffractometer	Bruker AXS 1000 CCD	Temperature, K	223(2)
μ(Mo-Kα), mm ⁻¹	0.28	<i>D</i> _{calcd} , g cm ⁻³	1.187
<i>F</i> (000)	624	θ _{max} , °	23.28
Reflns meas.	6794	Reflns unique	1156
Reflns with <i>I</i> > 2σ(<i>I</i>)	903	<i>R</i> (<i>F</i> ²), <i>R</i> _w (<i>F</i> ²) (all data)	0.050, 0.121
ρ, e Å ⁻³	0.28	G.O.F	1.062
No. obs/No. para	1156/88		
Programs used	SHELX-97 [3], Ortep-3 [4]		
Deposition number	CCDC 159330		

Acknowledgements

We gratefully acknowledge financial support from the EPSRC (MLC)

References

- [1] C. Jones, G.A. Koutsantonis, C.L. Raston, *Polyhedron* (1993) **12** 1829.
- [2] B. Luo, V.G. Young, W.L. Gladfelter, *Chem. Commun.* (1999) 123.
- [3] G.M. Sheldrick, *SHELX-97*, University of Göttingen, Germany (1997).
- [4] L.J. Farrugia, *Ortep-3 for Windows*, University of Glasgow (1998).